

Alternative Buffer Materials Project

Comparison of the available buffer materials: advantages and disadvantages related to their hydraulic, mechanical and geochemical performance

This presentation

- Objectives of the experiment
- Experiment design
- Results from measurements in field (Test 1)
- Termination of test package 1
- Chemical-mineralogical characterization
- Summary of observations
- Outlook

Participants

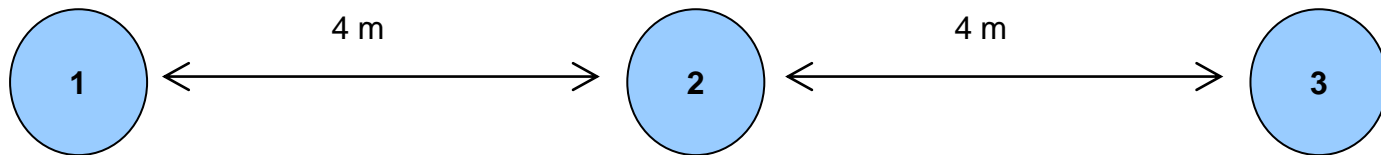
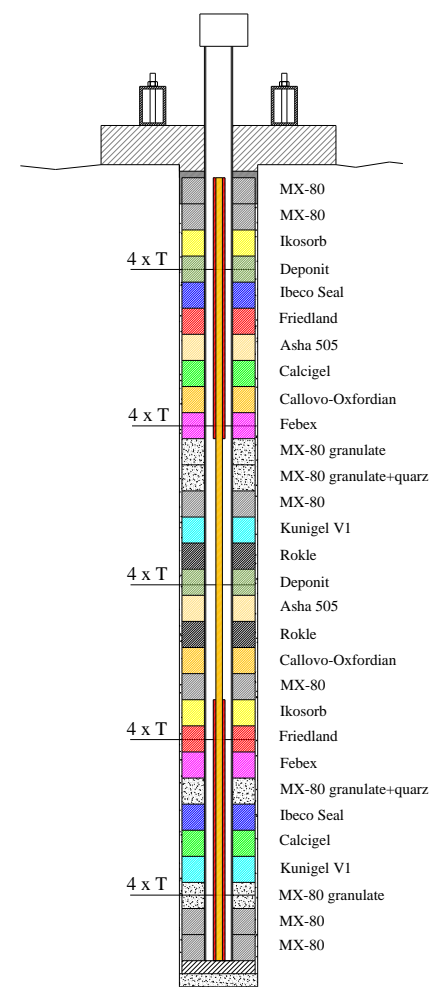
- SKB
- BGR
- Clay Technology
- Posiva
- Nagra/University of Bern
- ANDRA
- RAWRA
- JAEA
- AECL
- BRIUG
- KIT

Objectives of the project

- Detect and describe differences between different clay materials regarding:
 - Manufacturing and handling of blocks
 - Mineral stability
 - Physical properties
- Additional objectives:
 - Study of iron-clay interaction
 - Comparison of analytical methods
 - Diffusion of major cations

Experiment design

- Experiment design based on LOT with some modifications:
 - Different materials are tested
 - Less instrumentation
 - Shorter deposit holes (3m)
 - Complete package length heated
 - Central tube made of steel
 - No inserts or tracers



Short-term (1-2 years).
Heated from start.
Artificial saturation from start.

Medium-term(2-4 years).
Heating started when buffer was saturated.
Artificial saturation from start.

Long-term (≥ 5 years).
Heated from start.
No artificial saturation.

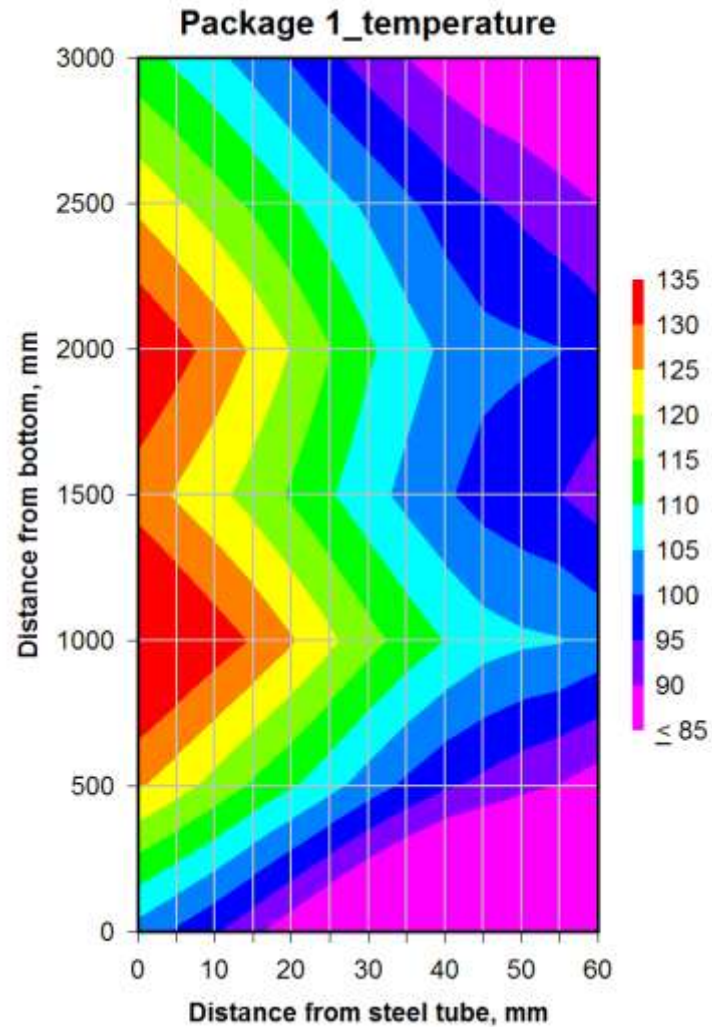
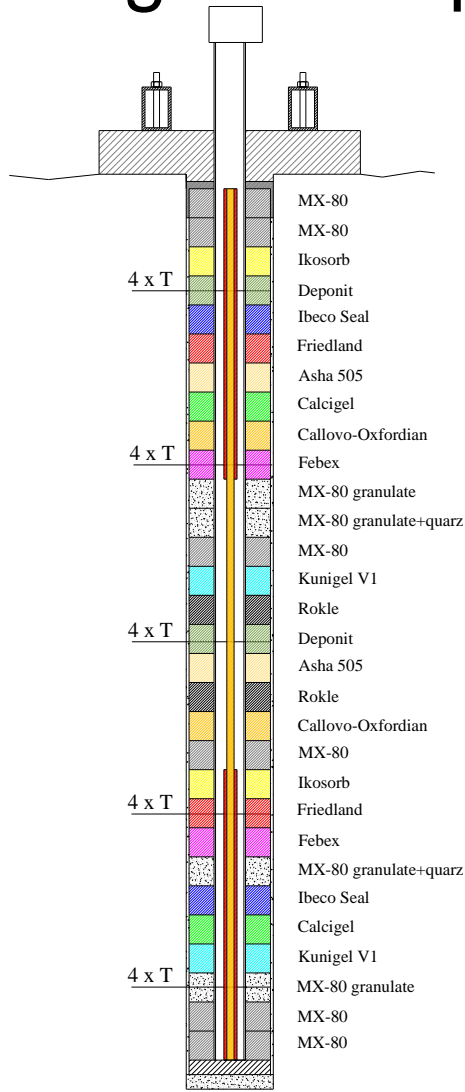
Materials



Standard investigation of reference materials

Material	Water content	Free swelling	Liquid limit	Grain density
	%	ml	%	kg/m ³
MX-80	12.9	17.0	545	2735
Calcigel	8.2	4.0	119	2695
Ikosorb	15.6	6.2	326	2740
Rokle	11.5	3.0	116	2940
Kunigel	7.8	9.7	462	2681
Febex	14.7	2.4	109	2735
Callovo Oxfordian	2.4	1.6	-	2682
Asha 505	12.7	8.8	337	2869
Friedland	5.2	3.9	68	2828
Ibeco Seal	14.0	11.2	522	2753
Deponit CAN	16.8	6.2	160	2678

Package 1 Temperature distribution



Termination of test package 1

- Seam drilling
- 2 x core drilled holes (d=300 mm)



Termination of test package 1



Termination of test package 1



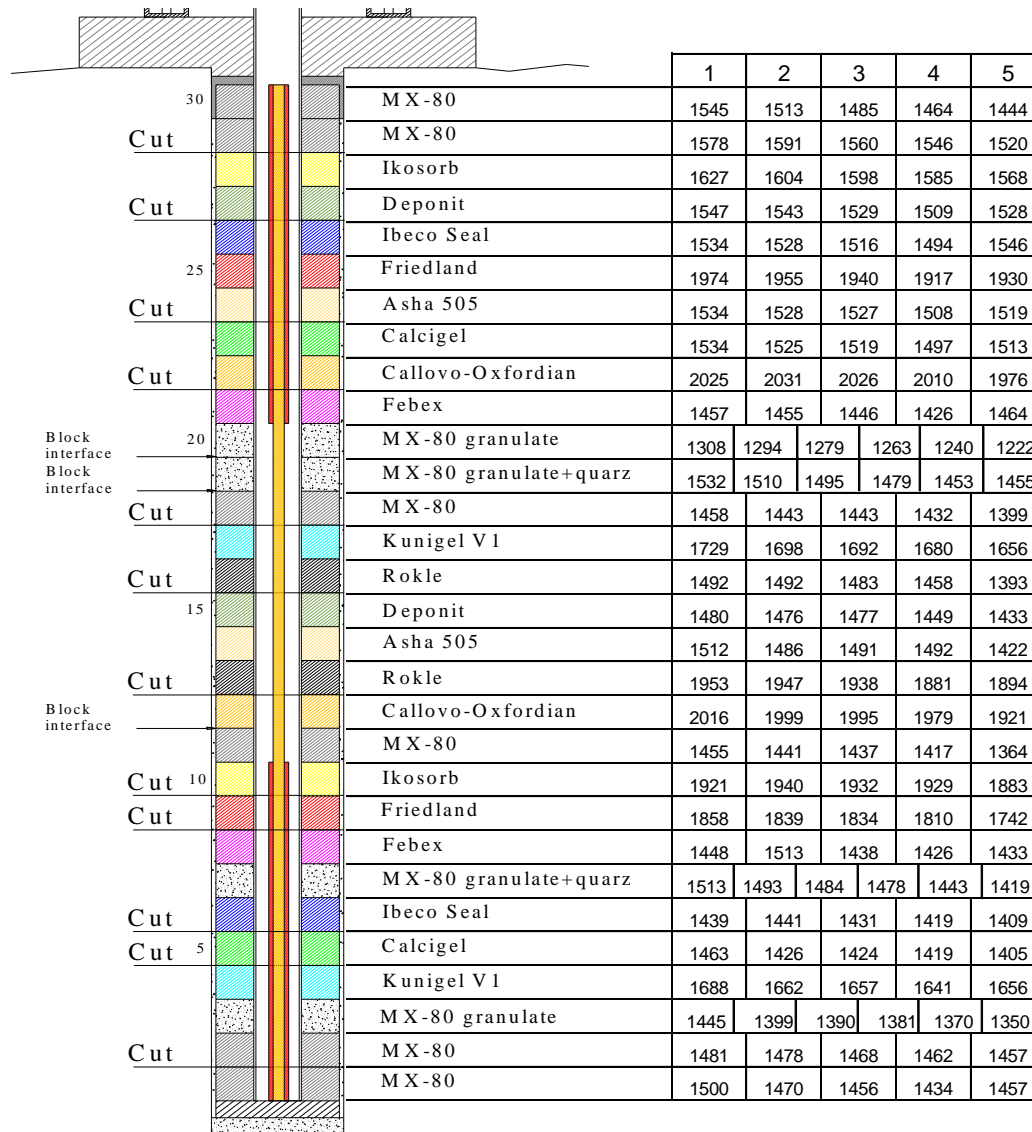
Termination of test package 1



Block division in laboratory



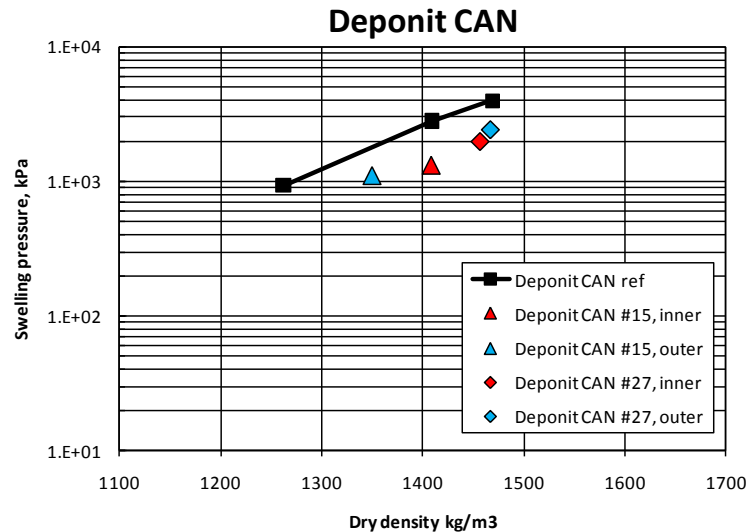
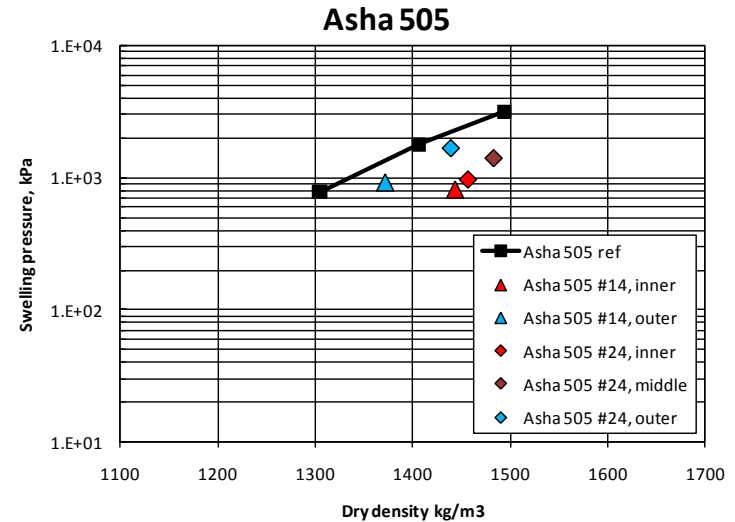
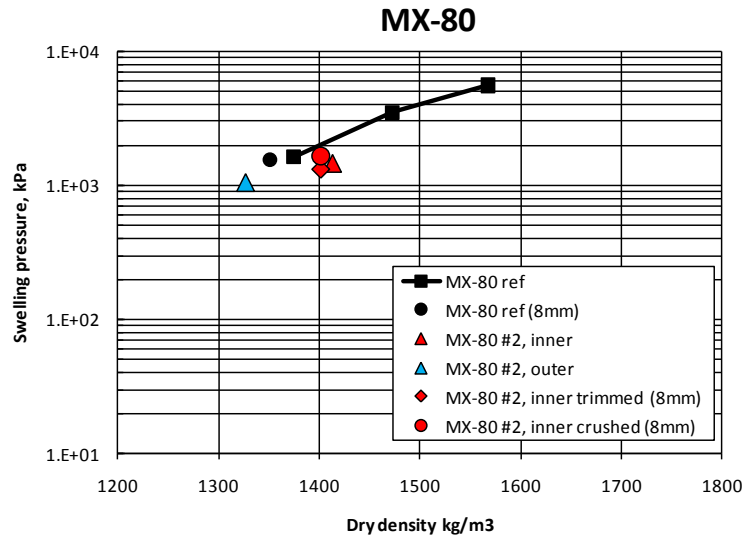
Density distribution



Swelling pressure and hydraulic conductivity












Swelling pressure



- There is a tendency that the swelling pressure of the test material is lower than the reference material
- The amount of tests is limited

Sample set

	JNB ABM 'Kunigel V1' Bentonite #17	CAN ABM 'Deponit Ca N' Bentonite #15	ROK ABM 'RAWRA' = ROK???? #13	CAL ABM 'Caligel' Bentonite #23	IBE ABM 'IBECO Seal' Bentonite #6
<i>Block</i>	<i>"metallic" crust, hard, a little rusty but thin</i>	<i>blue until 1 cm depths, direct contact bluegreen</i>	<i>thin black crust at contact, pristine clay underneath</i>	<i>a thick (1mm) black layer</i>	<i>black layer at contact with white spots, smells rotten</i>
					
1 mm	JNB 0	CAN 0	ROK 0	CAL 0	IBE 0
1 cm	JNB 1	CAN 1	ROK 1	CAL 1	IBE 1
3 cm	JNB 2	CAN 2	ROK 2	CAL 2	IBE 2
5 cm	JNB 3	CAN 3	ROK 3	CAL 3	IBE 3
7 cm	JNB 4	CAN 4	ROK 4	CAL 4	IBE 4
					IBE 0-1 white (1mg)

	LOT ABM 'MX 80' Bentonite #11	ASH ABM 'Asha 505' Bentonite #14	FRI ABM 'Friedland' Clay #9	FEB ABM 'Febex' Bentonite #8	IKO ABM 'Ikosorb' Bentonite #10
<i>Block</i>	<i>total sample dark, contact is brittle</i>	<i>no colour change at contact, smells as a disinfectant</i>	<i>pieces fell apart, contact very hard</i>	<i>blue seems, contact brittle, thick, black</i>	<i>very hard dark corrosion products at contact</i>
					
1 mm	LOT 0	ASH 0	FRI 0	FEB 0	IKO 0
1 cm	LOT 1	ASH 1	FRI 1	FEB 1	IKO 1
3 cm	LOT 2	ASH 2	FRI 2	FEB 2	IKO 2
5 cm	LOT 3	ASH 3	FRI 3	FEB 3	IKO 3
7 cm	LOT 4	ASH 4	FRI 4	FEB 4	IKO 4

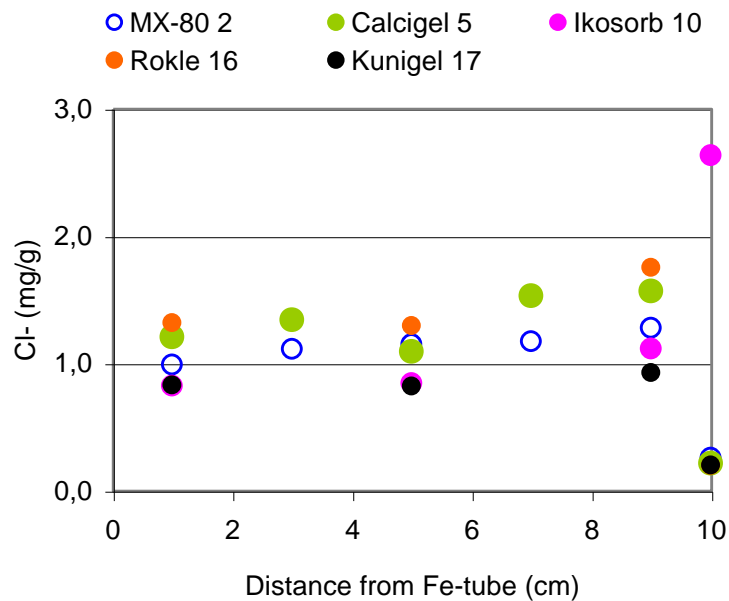
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74 Figure 2: Fotographs of the blocks before sampling and sample list.

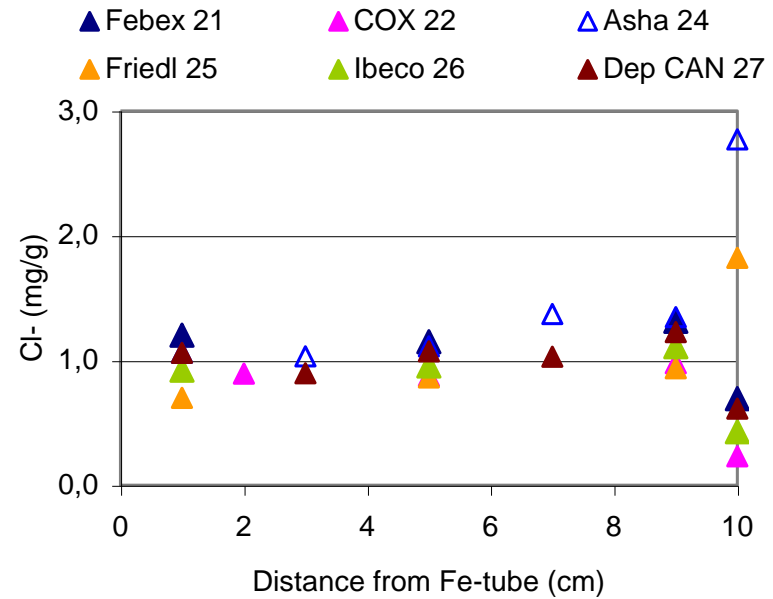


Water-soluble chloride (dispersion in water)

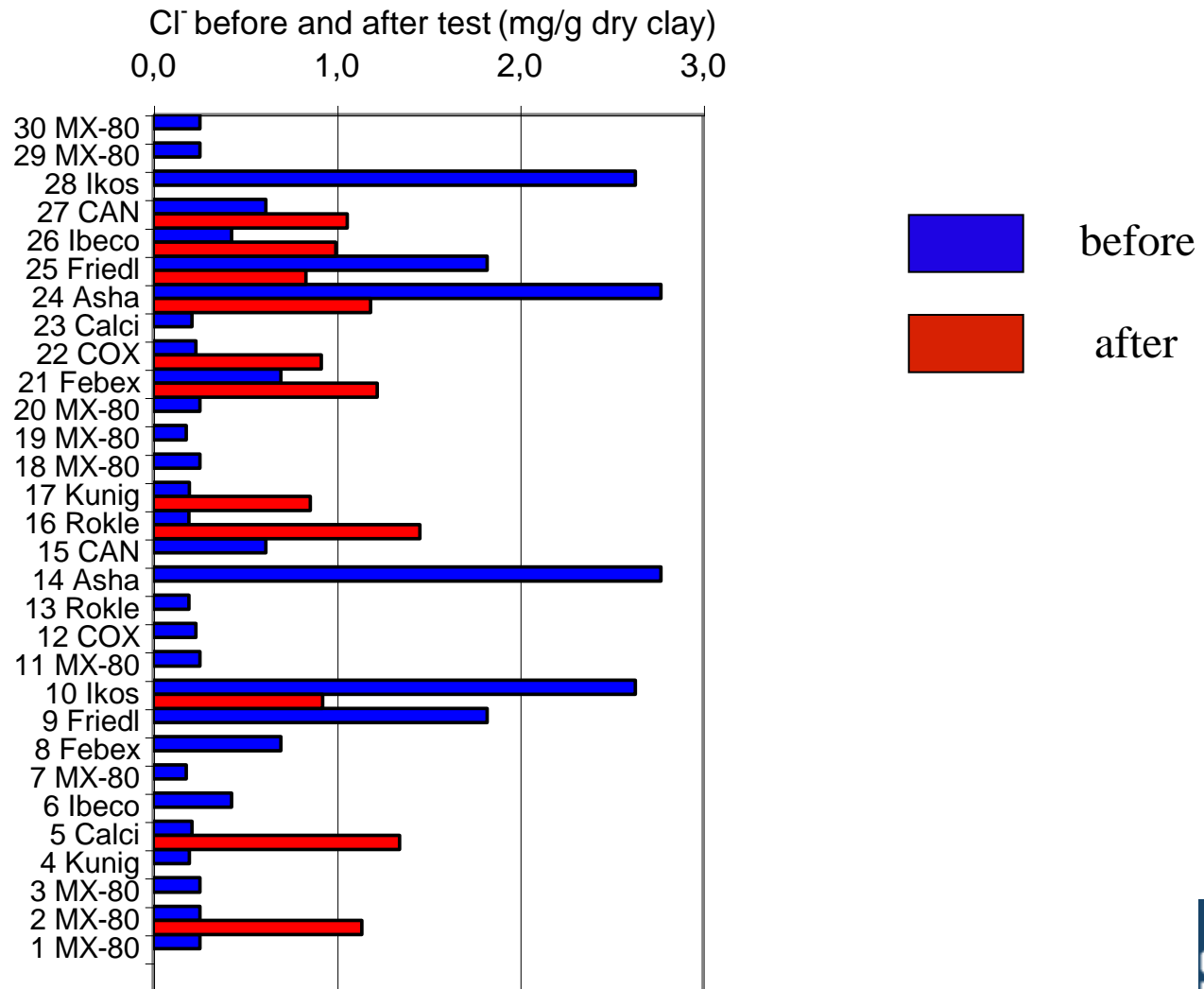
Lower part



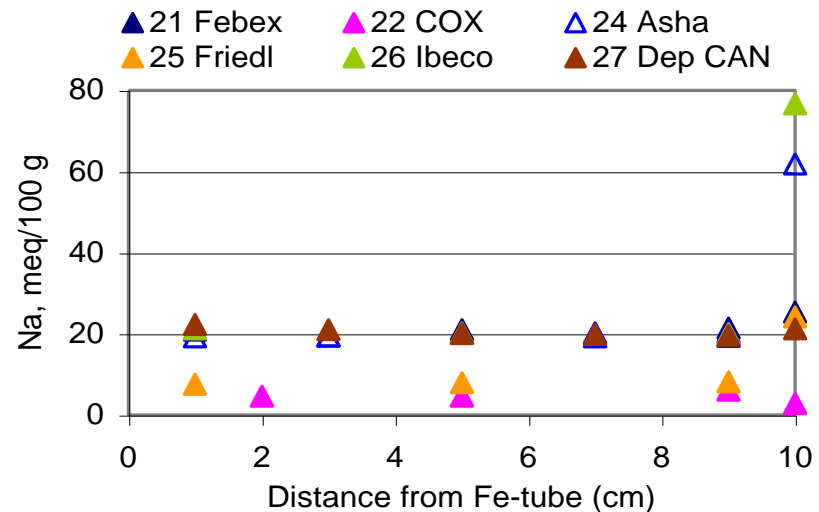
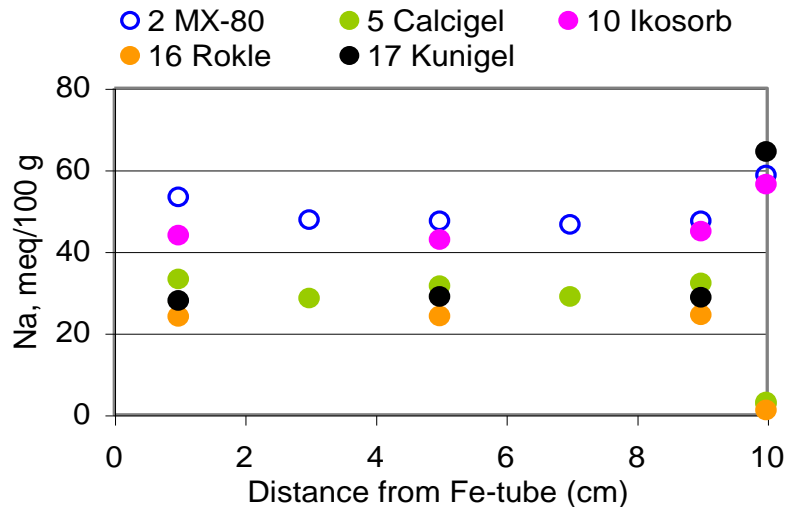
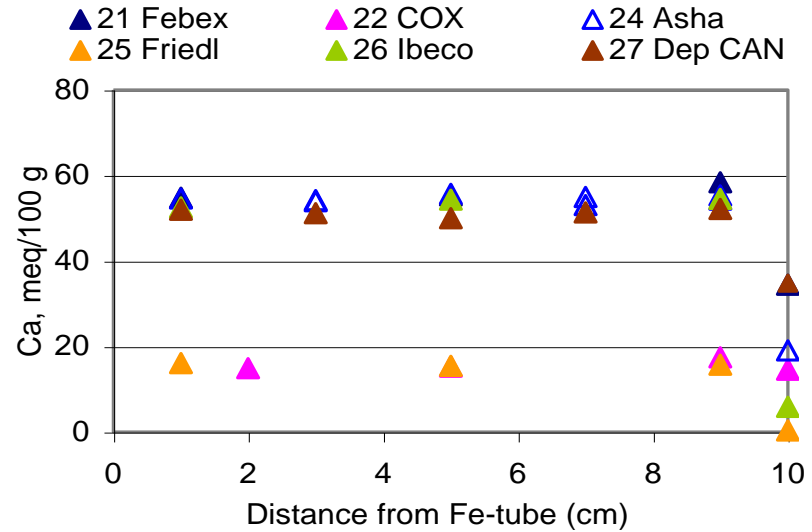
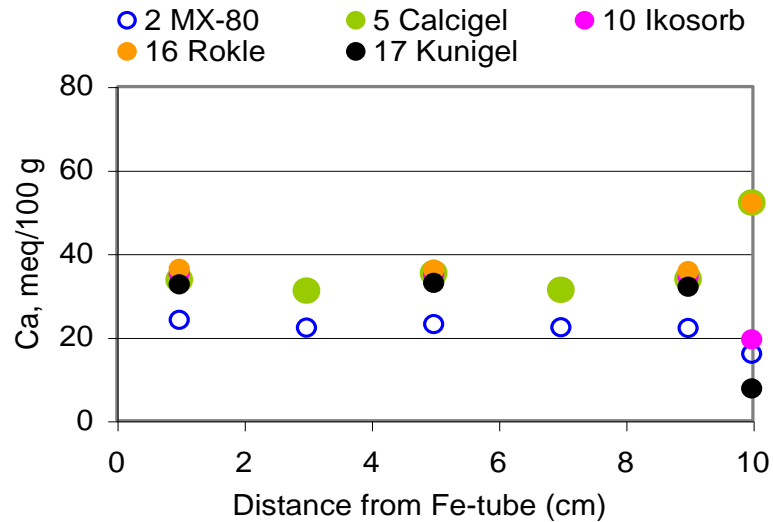
Upper part



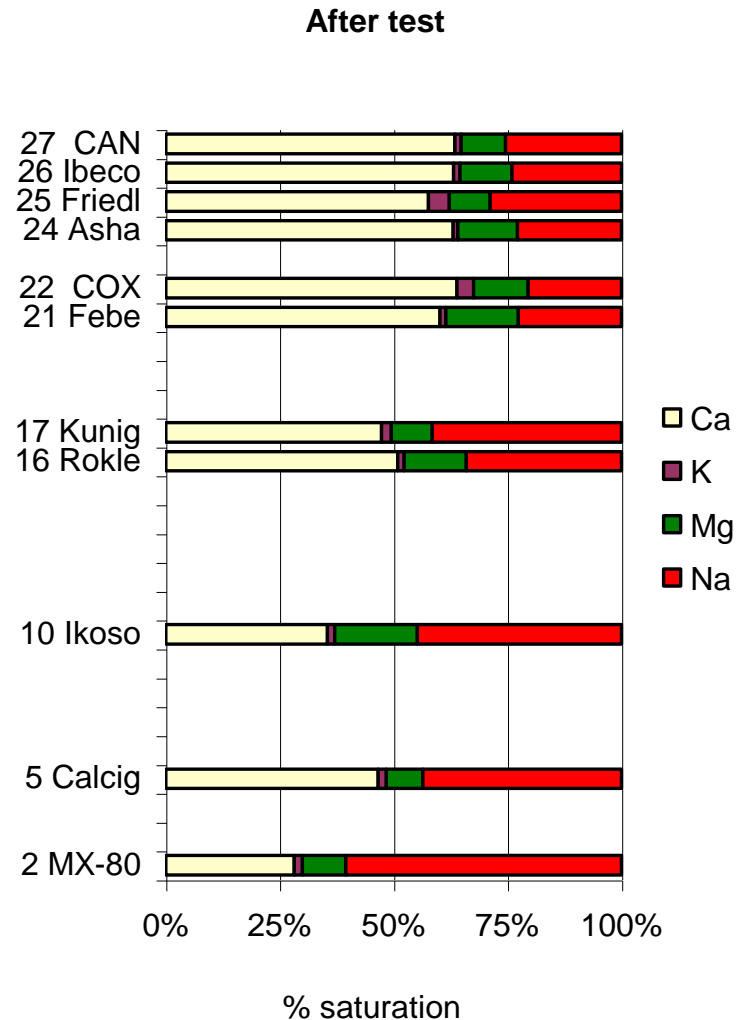
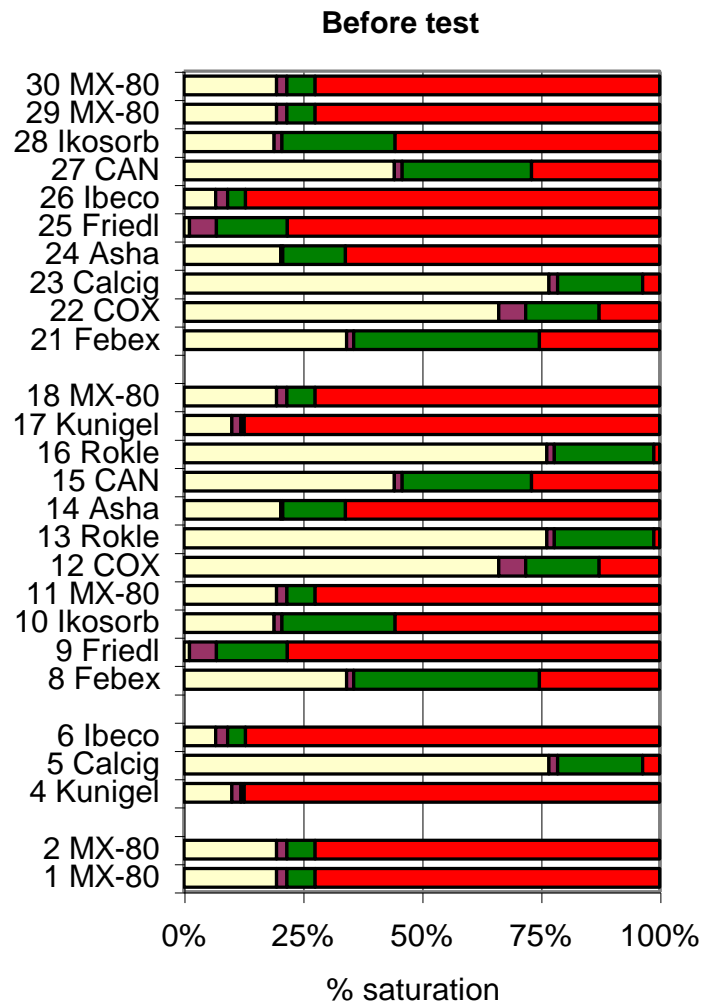
Chloride before and after the test



Exchangeable Ca and Na

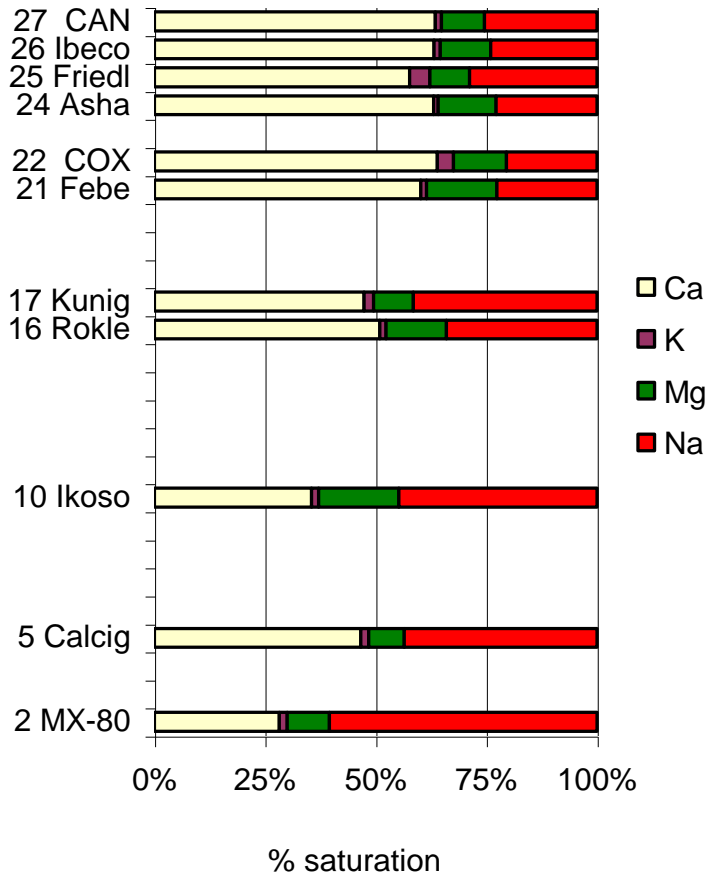


Relative cation distribution before and after the test

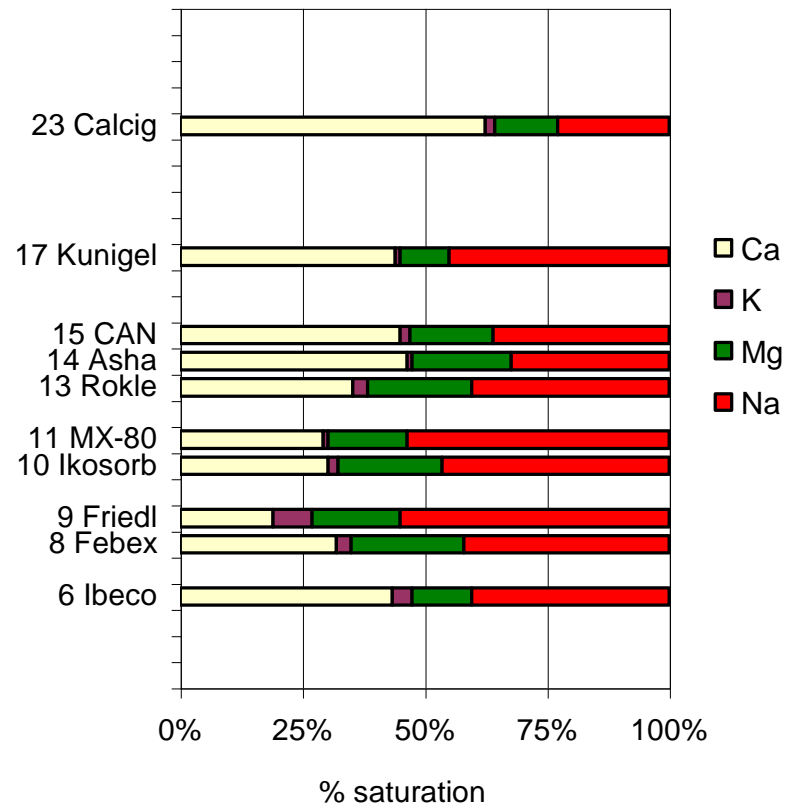


Comparison with BGR data

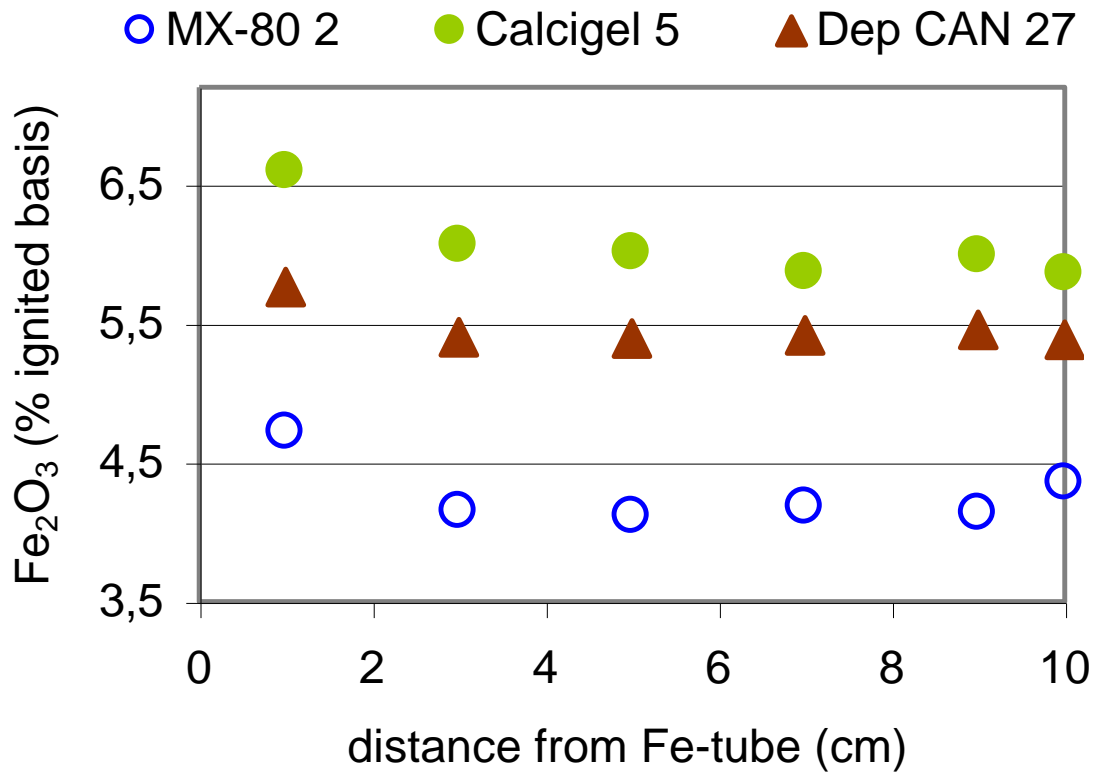
After test



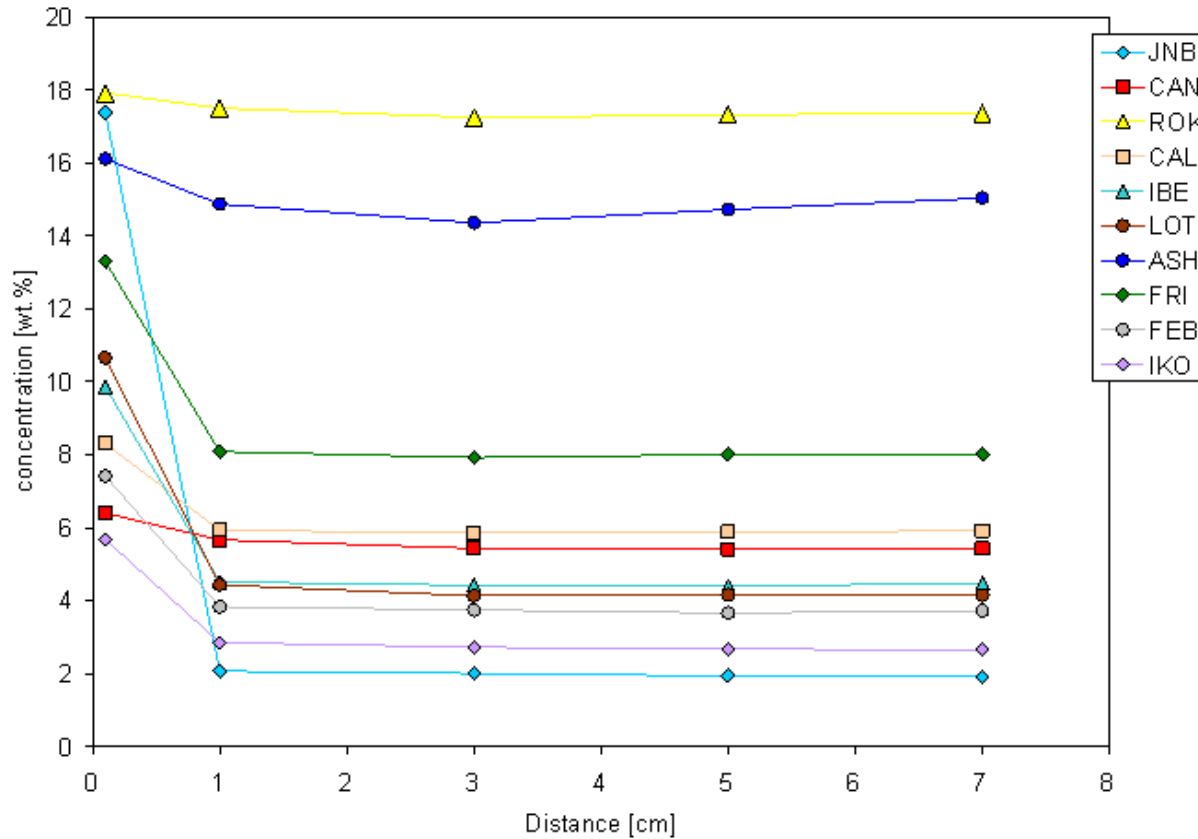
After test - BGR-data



Iron distribution, bulk samples (Clay Technology)



Iron distribution, bulk samples (BGR)

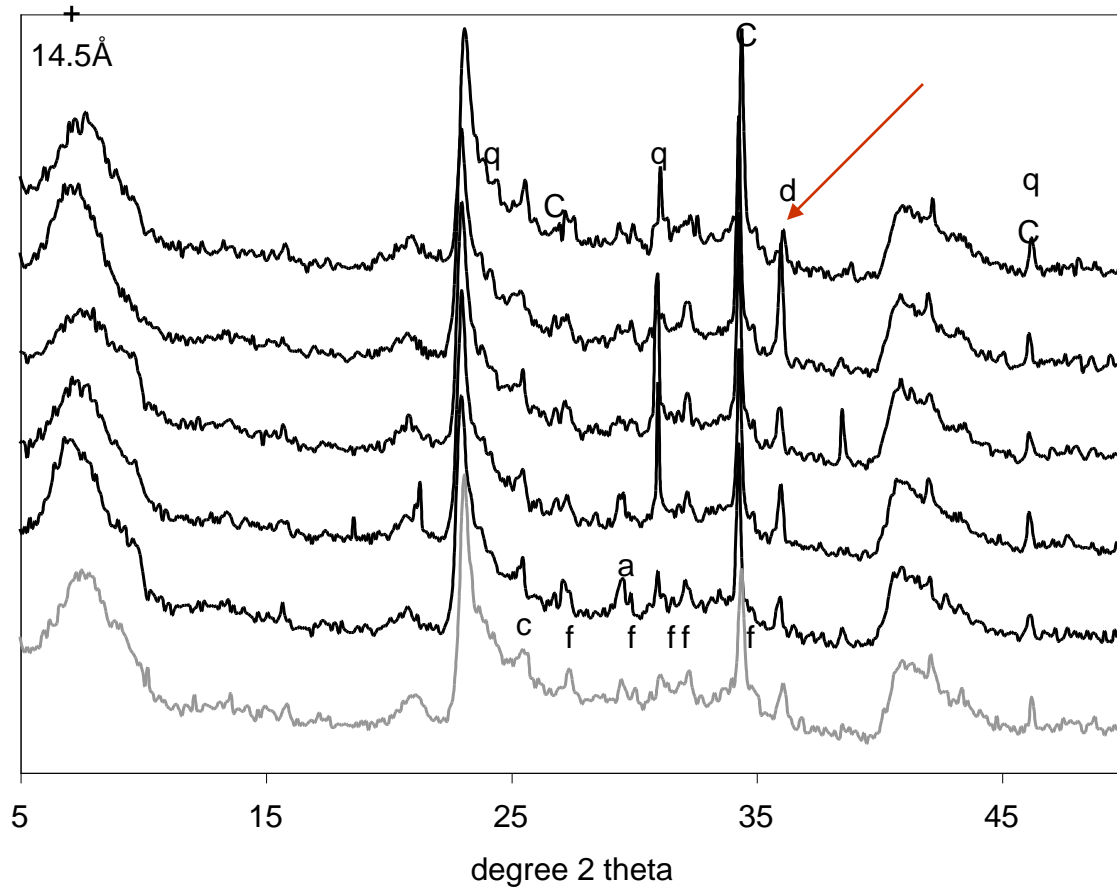


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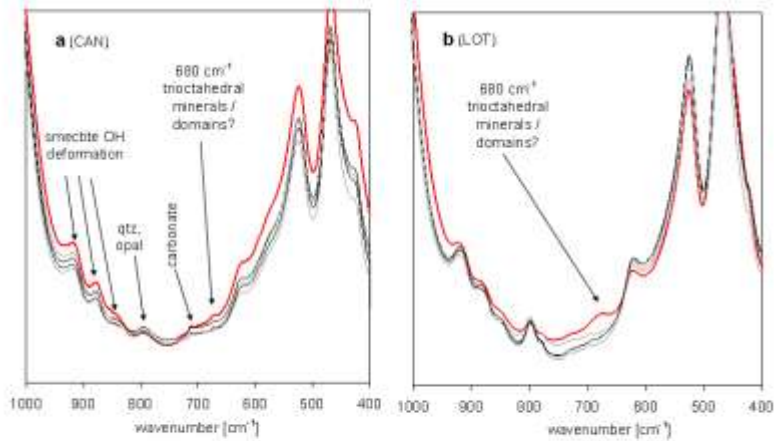
115 Figure 3: Content of Fe_2O_3 depending on the distance to the heater.

XRD-profiles of block 27, Deponit CAN-dolomite dissolution??

Block 27 Deponit CAN



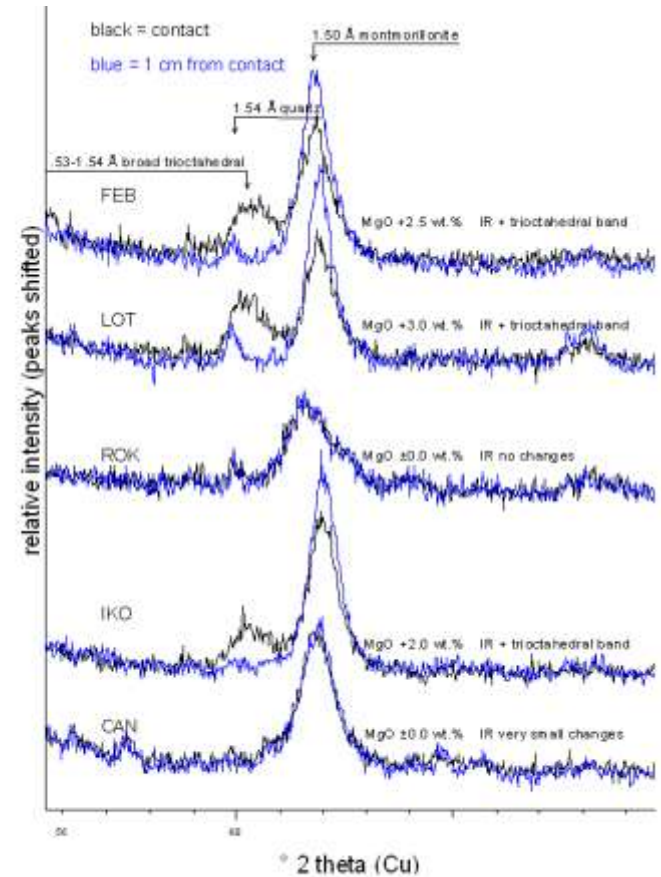
Trioctahedral minerals/domains



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250 Abb 13: IR spectra of samples CAN and LOT showing increased 680 cm⁻¹ intensity

251 which can not be attributed to precipitation of sulphates.



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260 Figure 14: XRD 060 reflection of selected samples.

Main observations

- The saturation with groundwater resulted in more or less constant chloride concentration in all blocks
- differential equilibration of the exchangeable cation pool in the upper and lower part of the package
- Iron has been incorporated in the bentonite closest to the steel tube
- All blocks are depleted in magnesium and the loss is at a maximum in the outer parts
- Carbonate dissolution increases with temperature
- Nature of the trioctahedral mineral / domains?

Outlook

- New packages will be installed during 2012
 - Large interest in the project
 - Similar setup, but some materials will be replaced
 - Will generate data in the far future
- ABM 2 will be excavated in the beginning of 2013
- Parts of the “sister test” concrete & clay installed
 - More packages in the new tunnel